

2-(1*H*-Benzimidazol-2-yl)-*N*-(*E*-(dimethylamino)methylidene]benzene-sulfonamideAdnan Ashraf,<sup>a</sup> M. Nawaz Tahir,<sup>b\*</sup> Waseeq Ahmad Siddiqui<sup>a</sup> and Nadia Perveen<sup>a</sup><sup>a</sup>University of Sargodha, Department of Chemistry, Sargodha, Pakistan, and<sup>b</sup>University of Sargodha, Department of Physics, Sargodha, Pakistan

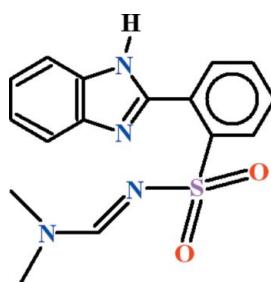
Correspondence e-mail: dmntahir\_uos@yahoo.com

Received 31 May 2012; accepted 2 June 2012

Key indicators: single-crystal X-ray study;  $T = 296$  K; mean  $\sigma(C-C) = 0.005$  Å;  $R$  factor = 0.059;  $wR$  factor = 0.159; data-to-parameter ratio = 15.1.

The asymmetric unit of the title compound,  $C_{16}H_{16}N_4O_2S$ , contains two molecules (*A* and *B*) with similar conformations: the benzene rings are oriented at dihedral angles of  $80.94(10)^\circ$  and  $84.54(10)^\circ$  with their adjacent  $1H$ -benzimidazole groups. In the crystal, the molecules are connected by  $N-H \cdots N$  hydrogen bonds, forming separate *C*(4) chains of both the *A* and *B* molecules along [010]. The *A* and *B* chains are cross-linked by several  $C-H \cdots O$  interactions involving the benzene rings and the sulfonyl groups, which lead to  $R_2^1(5)$  loops in the linkage of the chains.

## Related literature

For a related structure, see: Esparza-Ruiz *et al.* (2010).

## Experimental

## Crystal data

 $C_{16}H_{16}N_4O_2S$  $M_r = 328.39$ Monoclinic,  $P2_1/n$ 
 $a = 15.630(5)$  Å  
 $b = 10.003(4)$  Å  
 $c = 22.122(5)$  Å

 $\beta = 110.657(5)^\circ$   
 $V = 3236.3(18)$  Å<sup>3</sup>  
 $Z = 8$   
Mo  $K\alpha$  radiation

 $\mu = 0.22$  mm<sup>-1</sup>  
 $T = 296$  K  
 $0.28 \times 0.20 \times 0.18$  mm

## Data collection

Bruker Kappa APEXII CCD  
diffractometer  
Absorption correction: multi-scan  
(*SADABS*; Bruker, 2005)  
 $T_{\min} = 0.930$ ,  $T_{\max} = 0.952$ 

26702 measured reflections  
6347 independent reflections  
3328 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.070$ 

## Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.059$   
 $wR(F^2) = 0.159$   
 $S = 1.01$   
6347 reflections

419 parameters  
H-atom parameters constrained  
 $\Delta\rho_{\max} = 0.34$  e Å<sup>-3</sup>  
 $\Delta\rho_{\min} = -0.32$  e Å<sup>-3</sup>

**Table 1**  
Hydrogen-bond geometry (Å, °).

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
N1—H1 $\cdots$ N2 <sup>i</sup>	0.86	2.11	2.964 (3)	174
N5—H5A $\cdots$ N6 <sup>ii</sup>	0.86	2.10	2.955 (3)	178
C9—H9 $\cdots$ O4 <sup>i</sup>	0.93	2.56	3.153 (4)	122
C10—H10 $\cdots$ O4 <sup>i</sup>	0.93	2.57	3.153 (4)	121
C25—H25 $\cdots$ O2 <sup>ii</sup>	0.93	2.47	3.114 (5)	127

Symmetry codes: (i)  $-x + \frac{1}{2}, y - \frac{1}{2}, -z + \frac{1}{2}$ ; (ii)  $-x + \frac{3}{2}, y + \frac{1}{2}, -z + \frac{1}{2}$ .

Data collection: *APEX2* (Bruker, 2009); cell refinement: *SAINT* (Bruker, 2009); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 1997) and *PLATON* (Spek, 2009); software used to prepare material for publication: *WinGX* (Farrugia, 1999) and *PLATON*.

The authors acknowledge the provision of funds for the purchase of a diffractometer and encouragement by Dr Muhammad Akram Chaudhary, Vice Chancellor, University of Sargodha, Pakistan. The authors also acknowledge the technical support provided by Syed Muhammad Hussain Rizvi of Bana International, Karachi, Pakistan.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HB6832).

## References

Bruker (2005). *SADABS*. Bruker AXS Inc., Madison, Wisconsin, USA.  
Bruker (2009). *APEX2* and *SAINT*. Bruker AXS Inc., Madison, Wisconsin, USA.  
Esparza-Ruiz, A., Gonzalez-Gomez, G., Mijangos, E., Pena-Hueso, A., Lopez-Sandoval, H., Flores-Parra, A., Contreras, R. & Barba-Behrens, N. (2010). *Dalton Trans.* **39**, 6302–6309.  
Farrugia, L. J. (1997). *J. Appl. Cryst.* **30**, 565.  
Farrugia, L. J. (1999). *J. Appl. Cryst.* **32**, 837–838.  
Sheldrick, G. M. (2008). *Acta Cryst. A* **64**, 112–122.  
Spek, A. L. (2009). *Acta Cryst. D* **65**, 148–155.

## supplementary materials

*Acta Cryst.* (2012). **E68**, o2069 [doi:10.1107/S1600536812025159]

### **2-(1*H*-Benzimidazol-2-yl)-*N*-[(*E*)-(dimethylamino)methylidene]benzene-sulfonamide**

**Adnan Ashraf, M. Nawaz Tahir, Waseeq Ahmad Siddiqui and Nadia Perveen**

#### **Comment**

The title compound (I), (Fig. 1) has been prepared in an attempt to attach benzenesulfonyl chloride with 2-(1*H*-benzimidazol-2-yl) benzenesulfonamide (Crystal structure has been determined) in the dimethylformamide.

The crystal structures of 2-(1*H*-benzimidazol-3-ium-2-yl)benzenesulfonate dimethylsulfoxide solvate (Esparza-Ruiz *et al.*, 2010) has been published which is related to (I) upto some extent.

In (I), two molecules (M1 and M2) in the asymmetric unit are present, which differ slightly from each other geometrically. In molecule M1, the group A (C1—C7/N1/N2) of 1*H*-benzimidazole, benzene ring B (C8—C13) and group C (N3/C14/N4/C15/C16) of *N,N*-dimethylimidoformamide moiety are planar with r. m. s. deviation of 0.0108 Å, 0.0046 Å and 0.0093 Å, respectively. The dihedral angle between A/B, A/C and B/C is 80.94 (10)°, 12.34 (4)° and 83.76 (18)°, respectively. The sulfonyl group D (O1/S1/O2) is of course planar. The dihedral angle between B/D and C/D is 70.86 (14)° and 53.88 (13)°, respectively. In second molecule M2, the similar groups E (C17—C23/N5/N6), F (C24—C29) and G (N7/C30/N8/C31/C32) are also planar with r. m. s. deviation of 0.0160 Å, 0.0054 Å and 0.0122 Å, respectively. The dihedral angle between E/F, E/G and F/G is 84.54 (10)°, 12.68 (8)° and 83.22 (20)°, respectively. In M2, dihedral angle between F/H and G/H is 69.47 (14)° and 54.53 (13)°, respectively where H (O3/S2/O4) is the sulfonyl group. Both molecules are interlinked with themselves with C (4) chains due to classical H–bonding of N—H···N type (Table 1, Fig. 2). These infinite one-dimensional chains exist along [010]. The polymeric chains are interlinked with each other through benzene ring and the sulfonyl groups due to H–bonding of C—H···O type in a different manner. There exist  $R_2^1(5)$  ring motif in the linkage of polymeric chains.

#### **Experimental**

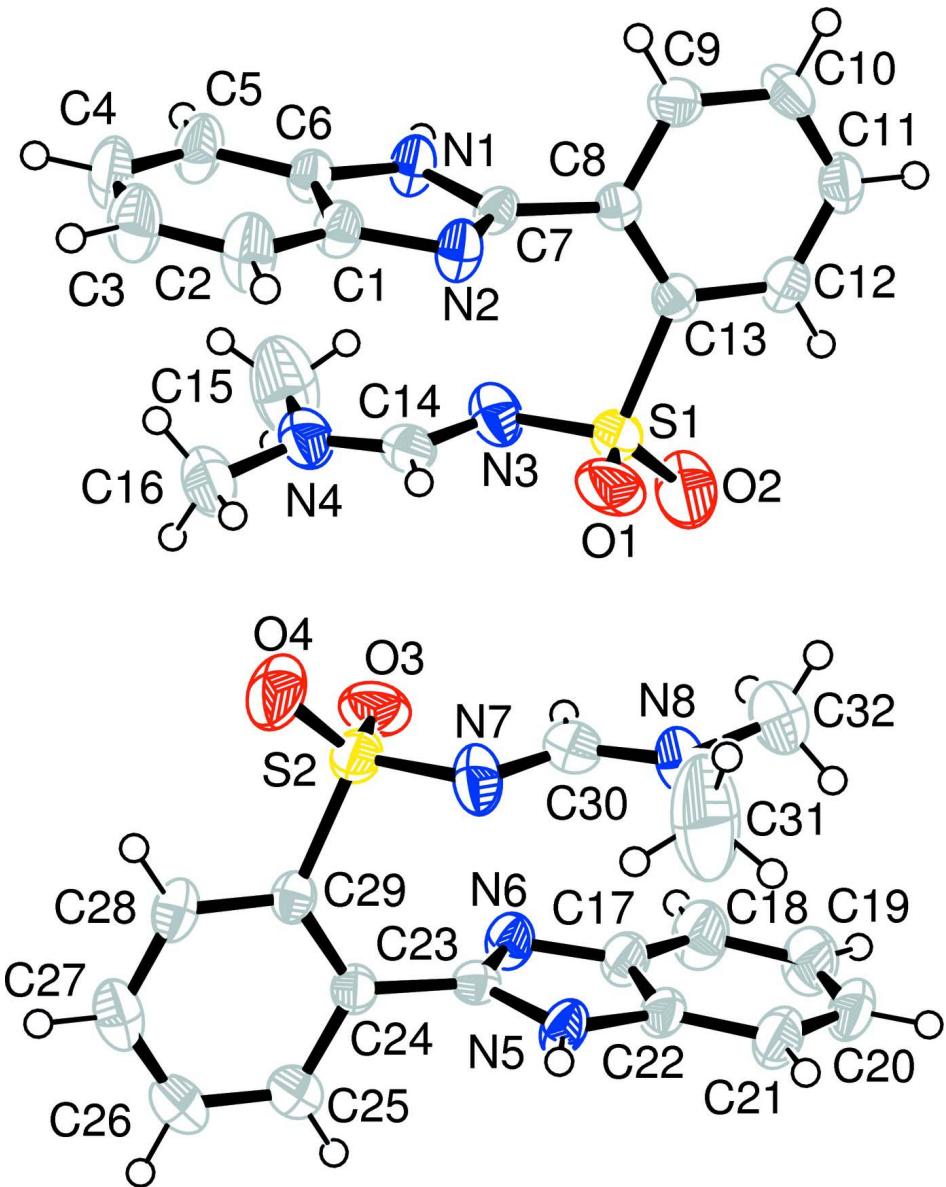
The 2-[*o*-(sulfamoyl)phenyl]benzimidazole (0.1 g, 0.37 mmol) in dimethylformamide (2 ml) was dissolved to get a clear solution. Benzenesulfonyl chloride (0.065 g, 0.37 mmol) was added with catalytic amount of potassium carbonate to this solution and subjected to reflux for 2 h. The resulting solution was quenched in ice-cold distilled water (100 ml). Extracted the aqueous layer with ethyl acetate (3 × 25 ml) and dried the organic layer over anhydrous sodium sulfate to get light brown powder (0.11 g, 90.5%). The powder was recrystallized in methanol to get the light brown prisms of (I). m.p. 593–594 K.

#### **Refinement**

The H-atoms were positioned geometrically at C—H = 0.93–0.96, and N—H = 0.86 Å, respectively and included in the refinement as riding with  $U_{\text{iso}}(\text{H}) = xU_{\text{eq}}(\text{C, N})$ , where  $x = 1.5$  for methyl H-atoms and  $x = 1.2$  for all other H-atoms.

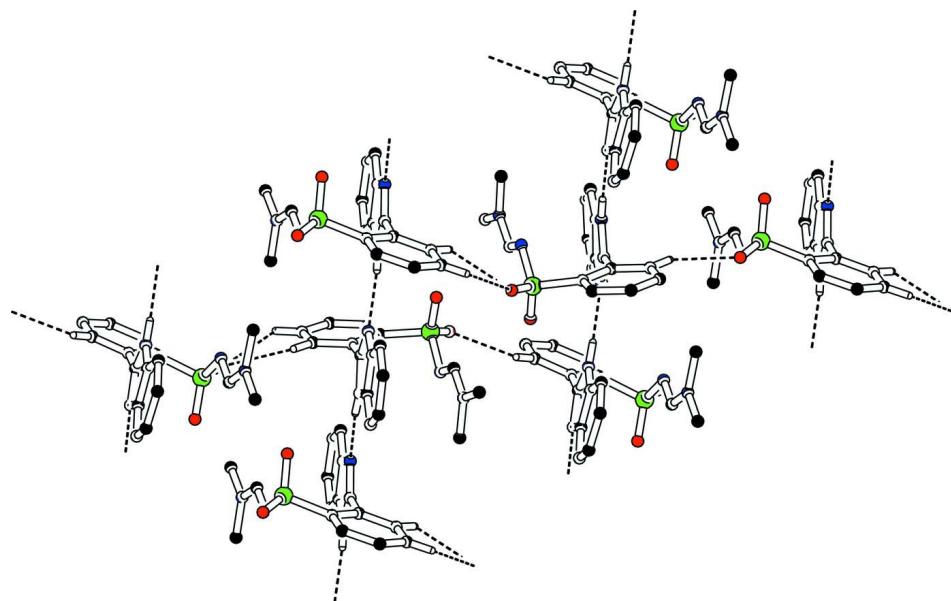
## Computing details

Data collection: *APEX2* (Bruker, 2009); cell refinement: *SAINT* (Bruker, 2009); data reduction: *SAINT* (Bruker, 2009); program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 1997) and *PLATON* (Spek, 2009); software used to prepare material for publication: *WinGX* (Farrugia, 1999) and *PLATON* (Spek, 2009).



**Figure 1**

View of the title compound with displacement ellipsoids drawn at the 50% probability level.

**Figure 2**

The partial packing (*PLATON*; Spek, 2009) which shows that molecules form C(4) chains and are interlinked. The H-atoms not involved in H-bondings are omitted for clarity.

### 2-(1H-Benzimidazol-2-yl)-N-[(E)- (dimethylamino)methylidene]benzenesulfonamide

#### Crystal data

$C_{16}H_{16}N_4O_2S$   
 $M_r = 328.39$   
 Monoclinic,  $P2_1/n$   
 Hall symbol: -P 2yn  
 $a = 15.630 (5) \text{ \AA}$   
 $b = 10.003 (4) \text{ \AA}$   
 $c = 22.122 (5) \text{ \AA}$   
 $\beta = 110.657 (5)^\circ$   
 $V = 3236.3 (18) \text{ \AA}^3$   
 $Z = 8$

$F(000) = 1376$   
 $D_x = 1.348 \text{ Mg m}^{-3}$   
 Mo  $K\alpha$  radiation,  $\lambda = 0.71073 \text{ \AA}$   
 Cell parameters from 3328 reflections  
 $\theta = 2.0\text{--}26.0^\circ$   
 $\mu = 0.22 \text{ mm}^{-1}$   
 $T = 296 \text{ K}$   
 Prism, light brown  
 $0.28 \times 0.20 \times 0.18 \text{ mm}$

#### Data collection

Bruker Kappa APEXII CCD  
 diffractometer  
 Radiation source: fine-focus sealed tube  
 Graphite monochromator  
 Detector resolution: 8.00 pixels  $\text{mm}^{-1}$   
 $\omega$  scans  
 Absorption correction: multi-scan  
 (*SADABS*; Bruker, 2005)  
 $T_{\min} = 0.930$ ,  $T_{\max} = 0.952$

26702 measured reflections  
 6347 independent reflections  
 3328 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.070$   
 $\theta_{\max} = 26.0^\circ$ ,  $\theta_{\min} = 2.0^\circ$   
 $h = -19 \rightarrow 19$   
 $k = -12 \rightarrow 12$   
 $l = -27 \rightarrow 22$

#### Refinement

Refinement on  $F^2$   
 Least-squares matrix: full  
 $R[F^2 > 2\sigma(F^2)] = 0.059$   
 $wR(F^2) = 0.159$   
 $S = 1.01$

6347 reflections  
 419 parameters  
 0 restraints  
 Primary atom site location: structure-invariant  
 direct methods

Secondary atom site location: difference Fourier map  
 Hydrogen site location: inferred from neighbouring sites  
 H-atom parameters constrained

$$w = 1/[\sigma^2(F_o^2) + (0.0638P)^2 + 0.4292P]$$

$$\text{where } P = (F_o^2 + 2F_c^2)/3$$

$$(\Delta/\sigma)_{\text{max}} < 0.001$$

$$\Delta\rho_{\text{max}} = 0.34 \text{ e \AA}^{-3}$$

$$\Delta\rho_{\text{min}} = -0.32 \text{ e \AA}^{-3}$$

#### Special details

**Geometry.** Bond distances, angles *etc.* have been calculated using the rounded fractional coordinates. All su's are estimated from the variances of the (full) variance-covariance matrix. The cell e.s.d.'s are taken into account in the estimation of distances, angles and torsion angles

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted  $R$ -factor  $wR$  and goodness of fit  $S$  are based on  $F^2$ , conventional  $R$ -factors  $R$  are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > \sigma(F^2)$  is used only for calculating  $R$ -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement.  $R$ -factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and  $R$ -factors based on ALL data will be even larger.

#### Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )

	$x$	$y$	$z$	$U_{\text{iso}}^*/U_{\text{eq}}$
S1	0.37734 (6)	0.20738 (11)	0.16157 (4)	0.0540 (4)
O1	0.38444 (16)	0.3454 (2)	0.18274 (13)	0.0766 (11)
O2	0.41107 (17)	0.1781 (4)	0.11124 (13)	0.1060 (16)
N1	0.25436 (18)	0.0481 (2)	0.27589 (11)	0.0378 (9)
N2	0.25592 (18)	0.2709 (2)	0.27357 (12)	0.0369 (9)
N3	0.42077 (18)	0.1030 (3)	0.21989 (13)	0.0507 (10)
N4	0.49756 (19)	0.0855 (3)	0.32963 (14)	0.0548 (11)
C1	0.2819 (2)	0.2276 (3)	0.33732 (15)	0.0362 (11)
C2	0.3058 (3)	0.2996 (3)	0.39450 (15)	0.0529 (14)
C3	0.3299 (3)	0.2312 (4)	0.45146 (16)	0.0563 (14)
C4	0.3314 (3)	0.0922 (3)	0.45294 (15)	0.0541 (16)
C5	0.3069 (3)	0.0181 (3)	0.39697 (15)	0.0510 (14)
C6	0.2813 (2)	0.0883 (3)	0.33941 (14)	0.0341 (11)
C7	0.24058 (19)	0.1606 (3)	0.23952 (14)	0.0293 (10)
C8	0.20535 (19)	0.1538 (3)	0.16793 (13)	0.0278 (9)
C9	0.1123 (2)	0.1313 (3)	0.13747 (15)	0.0398 (11)
C10	0.0741 (2)	0.1237 (3)	0.07082 (16)	0.0479 (11)
C11	0.1275 (2)	0.1359 (3)	0.03388 (15)	0.0429 (11)
C12	0.2199 (2)	0.1569 (3)	0.06293 (15)	0.0388 (11)
C13	0.25898 (19)	0.1670 (3)	0.12983 (14)	0.0304 (10)
C14	0.4588 (2)	0.1557 (3)	0.27730 (16)	0.0472 (11)
C15	0.4997 (3)	-0.0594 (4)	0.3281 (2)	0.1017 (19)
C16	0.5413 (3)	0.1492 (4)	0.39179 (17)	0.0774 (16)
S2	0.62238 (5)	0.50313 (10)	0.33544 (4)	0.0466 (3)
O3	0.62214 (15)	0.3695 (2)	0.31038 (12)	0.0628 (9)
O4	0.58453 (16)	0.5169 (3)	0.38516 (12)	0.0784 (13)
N5	0.73966 (18)	0.6823 (2)	0.22561 (11)	0.0399 (9)
N6	0.75783 (18)	0.4613 (2)	0.22988 (12)	0.0400 (9)
N7	0.57612 (18)	0.6125 (3)	0.28025 (13)	0.0495 (10)
N8	0.5008 (2)	0.6440 (4)	0.17119 (14)	0.0673 (13)
C17	0.7302 (2)	0.5032 (3)	0.16597 (15)	0.0412 (11)
C18	0.7158 (3)	0.4298 (4)	0.10993 (16)	0.0602 (14)
C19	0.6876 (3)	0.4976 (4)	0.05251 (17)	0.0652 (16)

C20	0.6720 (3)	0.6343 (4)	0.04889 (16)	0.0612 (16)
C21	0.6871 (3)	0.7089 (3)	0.10395 (16)	0.0539 (14)
C22	0.7180 (2)	0.6412 (3)	0.16255 (14)	0.0382 (11)
C23	0.76194 (19)	0.5718 (3)	0.26304 (14)	0.0309 (10)
C24	0.7940 (2)	0.5793 (3)	0.33451 (14)	0.0326 (10)
C25	0.8858 (2)	0.6104 (3)	0.36720 (16)	0.0478 (11)
C26	0.9208 (2)	0.6157 (4)	0.43360 (17)	0.0591 (14)
C27	0.8658 (3)	0.5928 (3)	0.46862 (17)	0.0539 (12)
C28	0.7741 (2)	0.5633 (3)	0.43755 (15)	0.0415 (11)
C29	0.7384 (2)	0.5552 (3)	0.37065 (14)	0.0320 (10)
C30	0.5412 (2)	0.5667 (4)	0.22150 (17)	0.0555 (16)
C31	0.4949 (4)	0.7876 (5)	0.1769 (2)	0.122 (3)
C32	0.4600 (3)	0.5844 (5)	0.10749 (19)	0.108 (2)
H1	0.24759	-0.03292	0.26196	0.0453*
H2	0.30546	0.39255	0.39403	0.0639*
H3	0.34568	0.27844	0.49005	0.0673*
H4	0.34927	0.04845	0.49255	0.0651*
H5	0.30761	-0.07483	0.39780	0.0608*
H9	0.07525	0.12129	0.16212	0.0479*
H10	0.01148	0.11022	0.05105	0.0573*
H11	0.10149	0.12994	-0.01087	0.0516*
H12	0.25644	0.16446	0.03780	0.0467*
H14	0.45815	0.24824	0.28106	0.0564*
H15A	0.45358	-0.09097	0.28921	0.1522*
H15B	0.48820	-0.09457	0.36489	0.1522*
H15C	0.55872	-0.08863	0.32916	0.1522*
H16A	0.53553	0.24449	0.38678	0.1162*
H16B	0.60491	0.12542	0.40845	0.1162*
H16C	0.51255	0.11996	0.42130	0.1162*
H5A	0.73910	0.76310	0.23871	0.0478*
H18	0.72508	0.33784	0.11139	0.0719*
H19	0.67852	0.45024	0.01457	0.0784*
H20	0.65107	0.67604	0.00876	0.0736*
H21	0.67706	0.80070	0.10202	0.0649*
H25	0.92372	0.62779	0.34384	0.0575*
H26	0.98240	0.63487	0.45476	0.0711*
H27	0.88990	0.59712	0.51348	0.0645*
H28	0.73658	0.54898	0.46149	0.0496*
H30	0.54482	0.47545	0.21474	0.0667*
H31A	0.53722	0.81604	0.21800	0.1837*
H31B	0.43392	0.81161	0.17345	0.1837*
H31C	0.50972	0.83024	0.14296	0.1837*
H32A	0.48518	0.62559	0.07828	0.1617*
H32B	0.39504	0.59798	0.09207	0.1617*
H32C	0.47291	0.49031	0.11006	0.1617*

Atomic displacement parameters ( $\text{\AA}^2$ )

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
S1	0.0310 (5)	0.0841 (8)	0.0447 (6)	-0.0017 (5)	0.0108 (4)	0.0174 (5)

O1	0.0589 (17)	0.0562 (18)	0.087 (2)	-0.0297 (13)	-0.0085 (15)	0.0247 (14)
O2	0.0421 (16)	0.230 (4)	0.0547 (19)	0.022 (2)	0.0280 (14)	0.026 (2)
N1	0.0621 (18)	0.0204 (13)	0.0280 (15)	-0.0069 (12)	0.0123 (13)	-0.0025 (11)
N2	0.0582 (18)	0.0215 (14)	0.0280 (15)	-0.0024 (12)	0.0116 (13)	-0.0009 (11)
N3	0.0426 (16)	0.0581 (19)	0.0441 (19)	0.0099 (14)	0.0063 (14)	0.0020 (15)
N4	0.0478 (17)	0.058 (2)	0.0480 (19)	-0.0014 (15)	0.0037 (15)	0.0097 (16)
C1	0.0462 (19)	0.0310 (18)	0.0311 (19)	-0.0014 (14)	0.0133 (15)	-0.0018 (14)
C2	0.086 (3)	0.0307 (19)	0.036 (2)	-0.0013 (19)	0.014 (2)	-0.0063 (16)
C3	0.082 (3)	0.052 (2)	0.033 (2)	-0.001 (2)	0.018 (2)	-0.0100 (17)
C4	0.074 (3)	0.056 (3)	0.028 (2)	-0.002 (2)	0.0125 (18)	0.0081 (17)
C5	0.081 (3)	0.0332 (19)	0.035 (2)	-0.0088 (18)	0.0156 (19)	0.0047 (16)
C6	0.0458 (19)	0.0259 (18)	0.0288 (18)	-0.0067 (14)	0.0109 (15)	-0.0004 (14)
C7	0.0359 (17)	0.0244 (16)	0.0289 (17)	-0.0020 (14)	0.0132 (14)	0.0021 (14)
C8	0.0324 (16)	0.0200 (16)	0.0293 (17)	-0.0010 (13)	0.0089 (14)	-0.0007 (12)
C9	0.0370 (18)	0.045 (2)	0.039 (2)	-0.0078 (15)	0.0154 (16)	-0.0052 (15)
C10	0.0317 (18)	0.058 (2)	0.041 (2)	-0.0067 (16)	-0.0034 (16)	-0.0018 (17)
C11	0.050 (2)	0.043 (2)	0.0280 (19)	-0.0027 (16)	0.0042 (17)	0.0017 (15)
C12	0.049 (2)	0.0368 (19)	0.0322 (19)	-0.0013 (16)	0.0162 (16)	0.0034 (14)
C13	0.0319 (16)	0.0262 (16)	0.0317 (18)	0.0009 (13)	0.0095 (14)	0.0019 (13)
C14	0.0338 (18)	0.053 (2)	0.050 (2)	-0.0061 (16)	0.0089 (17)	0.0087 (18)
C15	0.126 (4)	0.066 (3)	0.082 (3)	0.028 (3)	-0.002 (3)	0.010 (2)
C16	0.077 (3)	0.087 (3)	0.046 (2)	-0.024 (2)	-0.006 (2)	0.009 (2)
S2	0.0311 (4)	0.0672 (7)	0.0420 (5)	0.0038 (4)	0.0134 (4)	0.0173 (5)
O3	0.0529 (15)	0.0439 (15)	0.0805 (19)	-0.0146 (12)	0.0099 (14)	0.0084 (13)
O4	0.0459 (15)	0.146 (3)	0.0526 (17)	0.0202 (16)	0.0291 (14)	0.0309 (16)
N5	0.0627 (18)	0.0266 (14)	0.0311 (15)	-0.0061 (13)	0.0176 (13)	0.0016 (12)
N6	0.0562 (18)	0.0334 (15)	0.0306 (15)	-0.0038 (13)	0.0155 (13)	-0.0043 (12)
N7	0.0475 (17)	0.0610 (19)	0.0343 (17)	0.0147 (14)	0.0072 (14)	0.0086 (14)
N8	0.0533 (19)	0.099 (3)	0.0369 (19)	-0.0115 (19)	0.0002 (15)	0.0157 (18)
C17	0.057 (2)	0.0347 (19)	0.0308 (19)	-0.0103 (16)	0.0141 (16)	-0.0033 (15)
C18	0.091 (3)	0.047 (2)	0.040 (2)	-0.010 (2)	0.020 (2)	-0.0105 (18)
C19	0.092 (3)	0.068 (3)	0.033 (2)	-0.021 (2)	0.019 (2)	-0.015 (2)
C20	0.082 (3)	0.070 (3)	0.028 (2)	-0.025 (2)	0.015 (2)	0.0067 (18)
C21	0.075 (3)	0.046 (2)	0.037 (2)	-0.0132 (19)	0.015 (2)	0.0075 (17)
C22	0.051 (2)	0.037 (2)	0.0264 (18)	-0.0136 (15)	0.0134 (15)	-0.0023 (14)
C23	0.0309 (16)	0.0298 (17)	0.0320 (18)	-0.0047 (14)	0.0112 (14)	-0.0010 (14)
C24	0.0362 (17)	0.0317 (18)	0.0284 (18)	-0.0015 (14)	0.0097 (14)	0.0006 (14)
C25	0.0394 (19)	0.059 (2)	0.042 (2)	-0.0086 (17)	0.0107 (17)	-0.0008 (17)
C26	0.041 (2)	0.079 (3)	0.046 (2)	-0.0135 (19)	0.0015 (19)	-0.003 (2)
C27	0.059 (2)	0.059 (2)	0.033 (2)	0.002 (2)	0.0029 (19)	-0.0040 (17)
C28	0.052 (2)	0.047 (2)	0.0290 (19)	0.0068 (16)	0.0185 (16)	0.0038 (15)
C29	0.0338 (17)	0.0313 (17)	0.0294 (18)	0.0046 (13)	0.0093 (14)	0.0053 (13)
C30	0.038 (2)	0.073 (3)	0.050 (3)	-0.0114 (18)	0.0088 (18)	0.012 (2)
C31	0.155 (6)	0.104 (4)	0.078 (4)	0.053 (4)	0.004 (4)	0.029 (3)
C32	0.093 (4)	0.164 (5)	0.043 (3)	-0.056 (3)	-0.005 (2)	0.011 (3)

Geometric parameters ( $\text{\AA}$ ,  $\text{\textit{\textdegree}}$ )

S1—O1	1.450 (2)	C4—H4	0.9300
S1—O2	1.420 (3)	C5—H5	0.9300
S1—N3	1.612 (3)	C9—H9	0.9300
S1—C13	1.778 (3)	C10—H10	0.9300
S2—O4	1.428 (3)	C11—H11	0.9300
S2—N7	1.609 (3)	C12—H12	0.9300
S2—C29	1.780 (3)	C14—H14	0.9300
S2—O3	1.447 (2)	C15—H15C	0.9600
N1—C6	1.377 (4)	C15—H15A	0.9600
N1—C7	1.356 (4)	C15—H15B	0.9600
N2—C7	1.310 (4)	C16—H16C	0.9600
N2—C1	1.392 (4)	C16—H16B	0.9600
N3—C14	1.308 (4)	C16—H16A	0.9600
N4—C14	1.306 (4)	C17—C22	1.392 (4)
N4—C16	1.450 (5)	C17—C18	1.389 (5)
N4—C15	1.451 (5)	C18—C19	1.369 (5)
N1—H1	0.8600	C19—C20	1.386 (6)
N5—C22	1.377 (4)	C20—C21	1.376 (5)
N5—C23	1.351 (4)	C21—C22	1.390 (4)
N6—C17	1.389 (4)	C23—C24	1.482 (4)
N6—C23	1.316 (4)	C24—C29	1.394 (5)
N7—C30	1.303 (5)	C24—C25	1.395 (5)
N8—C32	1.454 (5)	C25—C26	1.376 (5)
N8—C30	1.319 (5)	C26—C27	1.364 (6)
N8—C31	1.448 (6)	C27—C28	1.385 (6)
N5—H5A	0.8600	C28—C29	1.388 (4)
C1—C2	1.387 (4)	C18—H18	0.9300
C1—C6	1.394 (4)	C19—H19	0.9300
C2—C3	1.365 (5)	C20—H20	0.9300
C3—C4	1.391 (5)	C21—H21	0.9300
C4—C5	1.376 (4)	C25—H25	0.9300
C5—C6	1.384 (4)	C26—H26	0.9300
C7—C8	1.484 (4)	C27—H27	0.9300
C8—C9	1.389 (5)	C28—H28	0.9300
C8—C13	1.389 (4)	C30—H30	0.9300
C9—C10	1.384 (5)	C31—H31A	0.9600
C10—C11	1.364 (5)	C31—H31B	0.9600
C11—C12	1.374 (5)	C31—H31C	0.9600
C12—C13	1.391 (4)	C32—H32A	0.9600
C2—H2	0.9300	C32—H32B	0.9600
C3—H3	0.9300	C32—H32C	0.9600
O1—S1—O2	116.3 (2)	N3—C14—H14	118.00
O1—S1—N3	113.32 (16)	N4—C14—H14	118.00
O1—S1—C13	107.12 (16)	N4—C15—H15B	109.00
O2—S1—N3	109.02 (19)	N4—C15—H15A	109.00
O2—S1—C13	105.41 (16)	N4—C15—H15C	109.00
N3—S1—C13	104.75 (15)	H15B—C15—H15C	109.00

O3—S2—N7	113.45 (15)	H15A—C15—H15C	109.00
O3—S2—C29	107.61 (15)	H15A—C15—H15B	109.00
O4—S2—N7	108.77 (16)	H16A—C16—H16B	109.00
O4—S2—C29	105.65 (15)	H16B—C16—H16C	109.00
N7—S2—C29	104.41 (15)	N4—C16—H16B	109.00
O3—S2—O4	116.02 (17)	H16A—C16—H16C	109.00
C6—N1—C7	106.9 (2)	N4—C16—H16A	109.00
C1—N2—C7	104.4 (2)	N4—C16—H16C	110.00
S1—N3—C14	115.8 (2)	N6—C17—C22	110.0 (3)
C14—N4—C16	121.4 (3)	N6—C17—C18	130.0 (3)
C14—N4—C15	121.6 (3)	C18—C17—C22	120.0 (3)
C15—N4—C16	117.0 (3)	C17—C18—C19	117.7 (4)
C6—N1—H1	127.00	C18—C19—C20	122.3 (3)
C7—N1—H1	127.00	C19—C20—C21	120.8 (3)
C22—N5—C23	107.1 (2)	C20—C21—C22	117.1 (3)
C17—N6—C23	104.5 (2)	N5—C22—C21	132.8 (3)
S2—N7—C30	115.9 (3)	N5—C22—C17	105.1 (2)
C30—N8—C32	119.6 (4)	C17—C22—C21	122.0 (3)
C31—N8—C32	118.2 (3)	N6—C23—C24	124.7 (3)
C30—N8—C31	122.3 (3)	N5—C23—C24	122.0 (3)
C23—N5—H5A	126.00	N5—C23—N6	113.2 (3)
C22—N5—H5A	126.00	C23—C24—C25	117.6 (3)
N2—C1—C6	110.0 (3)	C23—C24—C29	124.0 (3)
C2—C1—C6	119.4 (3)	C25—C24—C29	118.5 (3)
N2—C1—C2	130.6 (3)	C24—C25—C26	120.7 (3)
C1—C2—C3	118.6 (3)	C25—C26—C27	120.5 (3)
C2—C3—C4	121.3 (3)	C26—C27—C28	120.2 (3)
C3—C4—C5	121.3 (3)	C27—C28—C29	119.9 (3)
C4—C5—C6	116.9 (3)	S2—C29—C24	123.0 (2)
N1—C6—C5	132.5 (3)	S2—C29—C28	116.6 (2)
C1—C6—C5	122.3 (3)	C24—C29—C28	120.3 (3)
N1—C6—C1	105.1 (2)	N7—C30—N8	122.9 (4)
N1—C7—N2	113.6 (3)	C17—C18—H18	121.00
N2—C7—C8	125.2 (3)	C19—C18—H18	121.00
N1—C7—C8	121.1 (3)	C18—C19—H19	119.00
C9—C8—C13	118.3 (3)	C20—C19—H19	119.00
C7—C8—C13	124.6 (3)	C19—C20—H20	120.00
C7—C8—C9	117.1 (3)	C21—C20—H20	120.00
C8—C9—C10	120.7 (3)	C20—C21—H21	121.00
C9—C10—C11	120.5 (3)	C22—C21—H21	121.00
C10—C11—C12	119.9 (3)	C24—C25—H25	120.00
C11—C12—C13	120.2 (3)	C26—C25—H25	120.00
S1—C13—C12	116.2 (2)	C25—C26—H26	120.00
C8—C13—C12	120.4 (3)	C27—C26—H26	120.00
S1—C13—C8	123.3 (2)	C26—C27—H27	120.00
N3—C14—N4	123.6 (3)	C28—C27—H27	120.00
C3—C2—H2	121.00	C27—C28—H28	120.00
C1—C2—H2	121.00	C29—C28—H28	120.00
C2—C3—H3	119.00	N7—C30—H30	119.00

C4—C3—H3	119.00	N8—C30—H30	119.00
C3—C4—H4	119.00	N8—C31—H31A	109.00
C5—C4—H4	119.00	N8—C31—H31B	109.00
C4—C5—H5	122.00	N8—C31—H31C	109.00
C6—C5—H5	122.00	H31A—C31—H31B	110.00
C10—C9—H9	120.00	H31A—C31—H31C	109.00
C8—C9—H9	120.00	H31B—C31—H31C	109.00
C9—C10—H10	120.00	N8—C32—H32A	109.00
C11—C10—H10	120.00	N8—C32—H32B	109.00
C12—C11—H11	120.00	N8—C32—H32C	109.00
C10—C11—H11	120.00	H32A—C32—H32B	109.00
C11—C12—H12	120.00	H32A—C32—H32C	109.00
C13—C12—H12	120.00	H32B—C32—H32C	109.00
O1—S1—N3—C14	−2.2 (3)	C1—C2—C3—C4	0.4 (7)
O2—S1—N3—C14	129.0 (3)	C2—C3—C4—C5	−1.3 (8)
C13—S1—N3—C14	−118.6 (3)	C3—C4—C5—C6	0.4 (7)
O1—S1—C13—C8	−70.0 (3)	C4—C5—C6—N1	179.1 (4)
O1—S1—C13—C12	106.1 (3)	C4—C5—C6—C1	1.5 (6)
O2—S1—C13—C8	165.6 (3)	N2—C7—C8—C13	82.7 (4)
O2—S1—C13—C12	−18.3 (3)	N1—C7—C8—C9	77.7 (4)
N3—S1—C13—C8	50.6 (3)	N1—C7—C8—C13	−101.6 (4)
N3—S1—C13—C12	−133.3 (2)	N2—C7—C8—C9	−98.0 (4)
O4—S2—N7—C30	−127.7 (3)	C7—C8—C13—S1	−5.2 (4)
C29—S2—N7—C30	119.9 (3)	C7—C8—C13—C12	178.9 (3)
O3—S2—C29—C24	67.4 (3)	C13—C8—C9—C10	−0.7 (4)
O3—S2—C29—C28	−108.7 (3)	C7—C8—C9—C10	180.0 (3)
O4—S2—C29—C24	−168.1 (3)	C9—C8—C13—C12	−0.4 (4)
O4—S2—C29—C28	15.9 (3)	C9—C8—C13—S1	175.5 (2)
O3—S2—N7—C30	3.0 (3)	C8—C9—C10—C11	1.1 (5)
N7—S2—C29—C28	130.5 (2)	C9—C10—C11—C12	−0.5 (5)
N7—S2—C29—C24	−53.4 (3)	C10—C11—C12—C13	−0.6 (5)
C7—N1—C6—C5	−177.8 (4)	C11—C12—C13—C8	1.1 (5)
C6—N1—C7—N2	−0.1 (4)	C11—C12—C13—S1	−175.2 (2)
C7—N1—C6—C1	0.2 (4)	N6—C17—C18—C19	179.5 (4)
C6—N1—C7—C8	−176.2 (3)	C22—C17—C18—C19	−1.5 (6)
C7—N2—C1—C6	0.0 (4)	N6—C17—C22—N5	0.6 (4)
C7—N2—C1—C2	−179.5 (4)	N6—C17—C22—C21	−177.6 (4)
C1—N2—C7—N1	0.1 (4)	C18—C17—C22—N5	−178.6 (3)
C1—N2—C7—C8	176.0 (3)	C18—C17—C22—C21	3.3 (6)
S1—N3—C14—N4	−179.6 (3)	C17—C18—C19—C20	−1.0 (7)
C15—N4—C14—N3	−1.3 (6)	C18—C19—C20—C21	2.0 (8)
C16—N4—C14—N3	178.0 (4)	C19—C20—C21—C22	−0.3 (7)
C23—N5—C22—C21	177.1 (4)	C20—C21—C22—N5	−179.9 (4)
C22—N5—C23—N6	0.8 (4)	C20—C21—C22—C17	−2.3 (6)
C23—N5—C22—C17	−0.8 (4)	N5—C23—C24—C25	−81.3 (4)
C22—N5—C23—C24	176.6 (3)	N5—C23—C24—C29	99.8 (4)
C23—N6—C17—C22	−0.2 (4)	N6—C23—C24—C25	94.1 (4)
C17—N6—C23—N5	−0.4 (4)	N6—C23—C24—C29	−84.9 (4)

C23—N6—C17—C18	178.9 (4)	C23—C24—C25—C26	−178.3 (3)
C17—N6—C23—C24	−176.1 (3)	C29—C24—C25—C26	0.7 (5)
S2—N7—C30—N8	179.8 (3)	C23—C24—C29—S2	3.5 (4)
C31—N8—C30—N7	2.3 (6)	C23—C24—C29—C28	179.5 (3)
C32—N8—C30—N7	−177.3 (4)	C25—C24—C29—S2	−175.5 (2)
C6—C1—C2—C3	1.3 (6)	C25—C24—C29—C28	0.5 (4)
N2—C1—C6—N1	−0.1 (4)	C24—C25—C26—C27	−1.1 (5)
N2—C1—C2—C3	−179.2 (4)	C25—C26—C27—C28	0.3 (5)
N2—C1—C6—C5	178.1 (4)	C26—C27—C28—C29	0.9 (5)
C2—C1—C6—N1	179.5 (3)	C27—C28—C29—S2	174.9 (2)
C2—C1—C6—C5	−2.3 (6)	C27—C28—C29—C24	−1.3 (5)

Hydrogen-bond geometry ( $\text{\AA}$ , °)

$D—H\cdots A$	$D—H$	$H\cdots A$	$D\cdots A$	$D—H\cdots A$
N1—H1 $\cdots$ N2 <sup>i</sup>	0.86	2.11	2.964 (3)	174
N5—H5A $\cdots$ N6 <sup>ii</sup>	0.86	2.10	2.955 (3)	178
C9—H9 $\cdots$ O4 <sup>i</sup>	0.93	2.56	3.153 (4)	122
C10—H10 $\cdots$ O4 <sup>i</sup>	0.93	2.57	3.153 (4)	121
C25—H25 $\cdots$ O2 <sup>ii</sup>	0.93	2.47	3.114 (5)	127

Symmetry codes: (i)  $-x+1/2, y-1/2, -z+1/2$ ; (ii)  $-x+3/2, y+1/2, -z+1/2$ .